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Part-3; Interference by
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pressure, P_0 is the atmospheric pressure, then if the difference is 'h' then lung pressure can be calculated as follows.. $P_1 = P_0 + h\rho g$.. Example. A man blows into one end of a U-tube containing water until the levels differ by 40.0 cm. if the atmospheric pressure is 1.01×10^5 ...

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$P V / T = \text{constant}$ so $V_1 = (76 \times 1.0 \times 10^{-3} \times 290) / 273 \times 72 = 1.12 \times 10^{-3} \text{ m}^3$

2. A mass of 1,200 cm³ of oxygen at 27°C and a pressure 1.2 atmosphere is compressed until its volume is 600 cm³ and its pressure is 3.0 atmosphere.

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